# **FORMULA SHEET**

#### **General formulas:**

Newton's 2<sup>nd</sup> law of motion.F = mKinetic energy. $E_k = \frac{1}{2}$ Gravitational energy. $E_g = m$ Angular velocity of circular motion, where T is the period of the motion . $\omega = \frac{2\pi}{T}$ Ideal gas law. R is the Gas constantpV = mDensity in mass per unit volume. $\rho = \frac{m}{V}$ Specific heat: Heat needed to heat an object by 1 degree Celsius. Units are  $\frac{J}{kg*K}$ . $C = \frac{Q}{m\Delta}$ The conversion of going from Celsius to Kelvin. It is important to note that negative<br/>temperatures do not exist on the Kelvin scale, while they do for the Celsius scale, so<br/>when calculating with absolute temperatures, use Kelvin. In relative calculations where<br/>you take a temperature difference, it doesn't matter since Kelvin and Celsius are theF = m

same scale, except they are shifted. The radius of a circle, where r is the radius (half the diameter) of the circle.

The radius of a circle, where r is the radius (half the diameter) of the circle. $s = 2\pi r$ The area of a circle. $A = \pi r^2$ The volume of a sphere. $V = \frac{4}{3}\pi r^3$ 

$$F = m * a$$

$$E_k = \frac{1}{2}mv^2$$

$$E_g = m * g * h$$

$$\omega = \frac{2\pi}{T}$$

$$pV = nRT$$

$$\rho = \frac{m}{V}$$

$$C = \frac{Q}{m\Delta T}$$

$$T_K = T_{\circ C} + 273,15$$

#### Constants

 $N = 6.022 * 10^{23}$  The number of molecules in a mole, called Avogadro's Constant.

 $R = 8.315 \frac{J}{mol * K}$ . The gas constant

# **Quantities & Units**

Mass	m	kg
Time	t	S
Volume	V	m <sup>3</sup>
Velocity	v	m/s
Density	ρ	kg.m- <sup>3</sup>
Force	F	Ν
Temperature	Т	К
Pressure	p or P	Ра
Flow	$\phi$	kg.m <sup>-2</sup> s <sup>-1</sup>
Diffusion coefficient	D	
or diameter		
Internal energy	U	
Heat	Q	
Work	W	Nm
Total energy	Ε	J
Area	Α	m <sup>2</sup>
Heat transfer	h	
coefficient		
Thermal conductivity	λ	
Specific heat	$C_p$	
Drag coefficient	$C_D$	
Thermal diffusivity	a	
Viscosity	η	
Fourier's number	Fo	
Mass transfer	k	
coefficient		

# **WEEK 1:**

The general balance equation. $\frac{d}{dt}$	= in - out + production
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#### **WEEK 2:**

Total energy balance	$\frac{dE}{dt} = \phi_{m,in} * \left\{ U + \frac{p}{\rho} + \frac{1}{2}v^2 + gh \right\}_{in} - \phi_{m,out} * \left\{ U + \frac{p}{\rho} + \frac{1}{2}v^2 + gh \right\}_{out}$
First law of Thermodynamics, where $\Delta W$ is the net work done on the system.	$\Delta U = \Delta Q + \Delta W$
The thermal energy balance in a steady state without energy change.	$0 = \phi_m(u_{in} - u_{out}) + \phi_q + \phi_m e_{fr}$
The mechanical energy balance.	$0 = \phi_m \left( \frac{(v_{in}^2 - v_{out}^2)}{2} + g(h_{in} - h_{out}) + \frac{(p_{in} - p_{out})}{\rho} + \phi_w - \phi_m E_{fr} \right)$
Bernoulli's equation: Neglects all friction and heat production. $h$ is height.	$\frac{p}{\rho} + \frac{v^2}{2} + gh = constant$
Bernoulli's Principle: The energy per unit volume before is the same as the energy per unit volume after.	$P_1 + \frac{1}{2}\rho v_1^2 + \rho g h_1 = P_2 + \frac{1}{2}\rho v_2^2 + \rho g h_2$

### **WEEK 3:**

Reynolds number, where $\rho_f$ is the density of the fluid, $v_r$ is the relative velocity, D is the diameter and $\mu$ is the viscosity of the fluid	$Re = \frac{\rho_f v_r D}{\mu}$
The drag force. $C_D$ is the drag coefficient, A is the frontal area, v is the relative velocity.	$F_D = C_D A * \frac{1}{2} \rho_f v_r^2$
Stokes' law: The drag force on a sphere with a low Reynolds number ( $Re < 1$ ).	$F_D = 3\pi D\mu v_r$

#### **WEEK 4:**

Fourier's law, the transfer of heat. $\lambda$ is the material conductivity, $\Delta x$ is the thickness, A is the area, $\Delta T$ is the difference in temperature.	$\phi_q = \lambda A * \frac{\Delta T}{\Delta x}$
Fick's law of diffusion, analogous to Fourier's law. D is the	
diffusion coefficient, A is the area and $\frac{dC_a}{dx}$ is the change in	$\phi_m = -D * A * \left(\frac{dc_A}{dx}\right)$
concentration over x.	

#### **WEEK 5:**

Newton's law of cooling. $h$ is the heat transfer coefficient.	$\phi_q = h \cdot A \cdot \Delta T$
Nusselt number. Used to make h dimensionless.	$Nu = \frac{D \cdot h}{\lambda}$
Mass transfer coefficient, where $Sh$ is the Sherwood number, analogous to Nusselt number. $\Delta x$ is the size of the object, also called <i>D</i> sometimes.	$k = Sh \cdot \frac{D}{\Delta x}$

#### **WEEK 6:**

Thermal diffusivity. $\lambda$ is thermal conductivity, $\rho$ is material density, ${\cal C}_p$ is specific heat.	$a = \frac{\lambda}{\rho \cdot C_p}$
Penetration depth. Only valid while penetration theory still	
holds, for $\sqrt{\pi at} < \frac{D}{2}$ , where D is the size of the sheet being	$x_p = \sqrt{\pi a t}$
penetrated by heat.	
Fourier number.	$Fo = \frac{at}{D^2}$
Nusselt number for penetration theory.	$Nu = \sqrt{\frac{1}{\pi Fo}}$

**GRAPHS**:





